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INFRARED AND RAMAN SPECTRA OF SULFUR-NITRIDE COMPLEXES OF NICKEL(II)
AND PALLADIUM(II)

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KEY WORDS: IR Spectra, Raman Spectra, Sulfur Nitride Complexes,
Normal Coordinate Analysis

ABSTRACT

The infrared and Raman spectra of *trans*-Ni(S₂N₂CH₃)₂ and Pd(S₃N)₂ were measured from 4000-200 cm⁻¹. The absorption bands were assigned by comparison to the sulfur nitride complexes of nickel(II), palladium(II) and platinum(II). Normal coordinate analyses on these complexes were carried on these data using molecular parameters taken from X-ray data. To aid in band assignments, isotope shift data on *trans*-⁶²Ni(S₂N₂CH₃)₂ have also been carried out.

INTRODUCTION

In a previous paper we have reported on the infrared and Raman spectra of sulfur nitride square planar complexes of the type $M(S_2N_2H)_2$ with M is nickel, palladium and platinum (1). It was shown by the normal coordinate analyses that the vibrational modes associated with the sulfur nitrogen rings were strongly coupled. This data was also used to support an infrared study of the $Pt(S_2N_2H)_2$, $Pt(S_2N_2H)(S_3N)$ and $Pt(S_3N)_2$ complexes (2). This paper reports on two other sulfur nitride complexes, *trans*- $Ni(S_2N_2CH_3)_2$ and $Pd(S_3N)_2$, and makes use of the normal coordinate analysis technique combined with isotope shift data on $^{62}Ni(S_2N_2CH_3)_2$ to support assignments.

EXPERIMENTAL

1. *trans*- $Ni(S_2N_2CH_3)_2$

Trans- $Ni(S_2N_2CH_3)_2$ was prepared by reacting $Ni(S_2N_2H)_2$ with KOH and methyl iodide in tetrahydrofuran (THF) (3). The $Ni(S_2N_2H)_2$ was prepared as previously described (4,5). The THF was removed on a rotary evaporator and the residue was extracted with hot benzene. The combined benzene extracts were placed on an acid washed alumina column and the first eluate (green) contained sulfur and S_4N_4 . This was followed by a red fraction, $Ni(S_2N_2H)(S_3N)$, and then after ca. 3 hr a green fraction, $Ni(S_2N_2CH_3)_2$. A blue fraction, $Ni(S_2N_2CH_3)(S_2N_2H)$ and a violet fraction, $Ni(S_2N_2H)_2$ were eluted with acetone. The blue and green

fractions were further purified by chromatography followed by recrystallization from benzene. $\text{Ni}(\text{S}_2\text{N}_2\text{CH}_3)_2$, mp 194°C (lit. mp $194^\circ\text{C}(3)$); $\text{Ni}(\text{S}_2\text{N}_2\text{CH}_3)(\text{S}_2\text{N}_2\text{H})$, mp 144°C (lit. mp $144^\circ\text{C}(3)$).

The $^{62}\text{Ni}(\text{S}_2\text{N}_2\text{CH}_3)_2$ was prepared from $^{62}\text{Ni}(\text{S}_2\text{N}_2\text{H})_2$, whose synthesis was previously described (1).

2. $\text{Pd}(\text{S}_3\text{N})_2$

This compound was prepared by reacting PdCl_2 and S_4N_4 in methanol (6). After removal of the solvent followed by washing with water to remove chloride ion the product was dried in *vacuo*. The dry solid was taken up in cyclohexane and chromatographed on a *silica gel-G* column. The eluted blue solution was evaporated to ca. 5 mL and after ca. 12 hr $\text{Pd}(\text{S}_3\text{N})_2$ precipitated, mp 178°C (lit. mp 179°C (6)).

3. Spectral measurements

The infrared of the compounds were recorded on a Perkin-Elmer FIS-3 far i.r. spectrophotometer from $410\text{--}33\text{ cm}^{-1}$ as nujol mulls on a polyethylene plate. The compounds were also run on a Beckman IR-12 i.r. spectrophotometer from $4000\text{--}200\text{ cm}^{-1}$ using the KBr pellet technique. Raman spectra were recorded on a Spex Model 1401 double monochromator equipped with an ITT FW-130 phototube.

NORMAL COORDINATE ANALYSES

1. *trans*- $\text{Ni}(\text{S}_2\text{N}_2\text{CH}_3)_2$

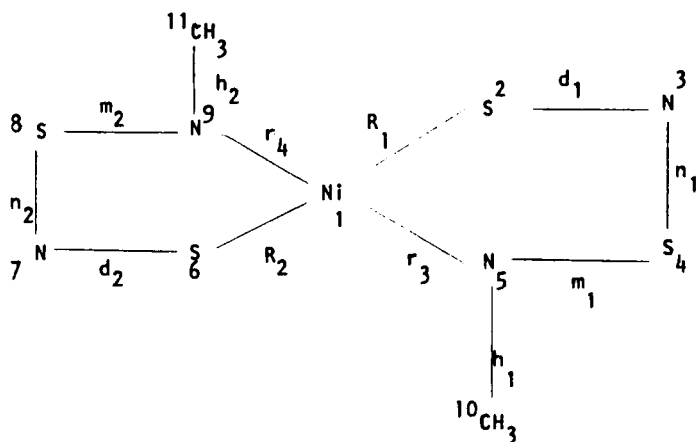
The normal coordinate analysis on this complex was carried out on a model of C_{2h} symmetry. The molecular parameters used

for this NCA were taken from the results of X-ray analysis (3). The twenty-seven normal vibrations can be classified into $9A_g + 3B_g + 5A_u + 10B_u$; the A_g and B_g are Raman active whereas the A_u and B_u modes are infrared active. In this treatment we have included all in-plane (A_g and B_u) vibrations. The programs written by Schachtschneider (7) were used in this work. The twenty-eight in-plane internal coordinates are shown in Table 1. The out-of-plane and torsion coordinates are not shown. These twenty-eight internal coordinates can be divided into twelve stretching and sixteen bending modes. The symmetry coordinates for *trans*- $Ni(S_2N_2CH_3)_2$ are shown in Table 2. Seven redundancies are included in the twenty-six symmetry coordinates. They were not easily eliminated and were included in the NCA calculations. Final results gave zero frequencies corresponding to these redundancies. The calculated force constants; stretching (K), bending (H) and repulsive (F) are displayed in Table 3. The interaction force constants (f) are not shown. Table 4 lists the observed and calculated spectra for *trans*- $Ni(S_2N_2CH_3)_2$ and its ^{62}Ni analog. Shown also are the potential energy distribution (PED) and the predominant modes.

2. $Pd(S_3N)_2$

The normal coordinate analysis on this planar complex was carried out on a model of C_{2v} symmetry. The molecular parameters used for the NCA were taken from X-ray analysis (8). The twenty-

TABLE I

Internal Coordinates for $\text{Ni}(\text{S}_2\text{N}_2\text{CH}_3)_2$ 

$$2-1-9 = \pi_1$$

$$6-1-5 = \pi_2$$

$$2-1-5 = \theta_1$$

$$6-1-9 = \theta_2$$

$$1-2-3 = \delta_1$$

$$1-6-7 = \delta_2$$

$$2-3-4 = \gamma_1$$

$$6-7-8 = \gamma_2$$

$$3-4-5 = \beta_1$$

$$7-8-9 = \beta_2$$

$$4-5-1 = \alpha_1$$

$$8-9-1 = \alpha_2$$

$$4-5-10 = \phi_1$$

$$8-9-11 = \phi_2$$

$$1-5-10 = \rho_1$$

$$1-9-11 = \rho_2$$

TABLE 2

Symmetry Coordinates for $\text{Ni}(\text{S}_2\text{N}_2\text{CH}_3)_2$

Ag: $S_1 = \Delta R_1 + \Delta R_2$	$S_8 = \Delta \pi_1 + \Delta \pi_2 - \Delta \theta_1 - \Delta \theta_2$
$S_2 = \Delta r_3 + \Delta r_4$	$S_9 = \Delta \delta_1 + \Delta \delta_2$
$S_3 = \Delta d_1 + \Delta d_2$	$S_{10} = \Delta \gamma_1 + \Delta \gamma_2$
$S_4 = \Delta m_1 + \Delta m_2$	$S_{11} = \Delta \beta_1 + \Delta \beta_2$
$S_5 = \Delta n_1 + \Delta n_2$	$S_{12} = \Delta \alpha_1 + \Delta \alpha_2 - \Delta \phi_1 - \Delta \phi_2 - \Delta \rho_1 - \Delta \rho_2$
$S_6 = \Delta h_1 + \Delta h_2$	$S_{13} = \Delta \phi_1 - \Delta \rho_2 - \Delta \rho_1 + \Delta \phi_2$
$S_7 = \Delta \pi_1 + \Delta \pi_2 + \Delta \theta_1 + \Delta \theta_2$	
Bu: $S_{14} = \Delta R_1 - \Delta R_2$	$S_{21} = \Delta \pi_1 - \Delta \pi_2 - \Delta \theta_1 + \Delta \theta_2$
$S_{15} = \Delta r_3 - \Delta r_4$	$S_{22} = \Delta \delta_1 - \Delta \delta_2$
$S_{16} = \Delta d_1 - \Delta d_2$	$S_{23} = \Delta \gamma_1 - \Delta \gamma_2$
$S_{17} = \Delta m_1 - \Delta m_2$	$S_{24} = \Delta \beta_1 - \Delta \beta_2$
$S_{18} = \Delta n_1 - \Delta n_2$	$S_{25} = \Delta \alpha_1 - \Delta \alpha_2 - \Delta \phi_1 + \Delta \phi_2 - \Delta \rho_1 + \Delta \rho_2$
$S_{19} = \Delta h_1 - \Delta h_2$	$S_{26} = \Delta \phi_1 + \Delta \rho_2 - \Delta \rho_1 - \Delta \phi_2$
$S_{20} = \Delta \pi_1 - \Delta \pi_2 + \Delta \theta_1 - \Delta \theta_2$	

TABLE 3

Force Constants (mdy/Å) for $\text{Ni}(\text{S}_2\text{N}_2\text{CH}_3)_2$

$K(\text{Ni-S})$	1.05	$H(\text{N-S-N})$	0.50
$K(\text{Ni-N})$	0.60	$H(\text{Ni-N-S})$	0.20
$K(\text{S}_1\text{-N}_1)$	5.40	$H(\text{S-N-C})$	0.35
$K(\text{S}_2\text{-N}_2)$	3.32	$H(\text{Ni-N-C})$	0.09
$K(\text{N}_1\text{-S}_2)$	2.65	$F(\text{S-S})$	0.40
$K(\text{N-C})$	6.20	$F(\text{N-N})$	0.40
$H(\text{S-Ni-N})$	0.15	$F(\text{Ni-S})$	0.40
$H(\text{S-Ni-N})$	0.13	$F(\text{S-C})$	0.55
$H(\text{Ni-S-N})$	0.30	$F(\text{Ni-C})$	0.40
$H(\text{S-N-S})$	0.80		

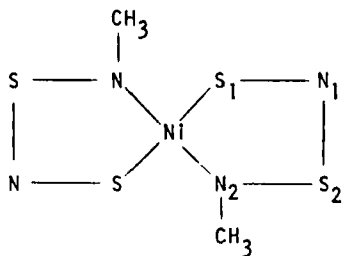


TABLE 4
Comparison of Observed and Calculated Frequencies and Band Assignments for $\text{Ni}(\text{S}_2\text{N}_2\text{CH}_3)_2(\text{cm}^{-1})$

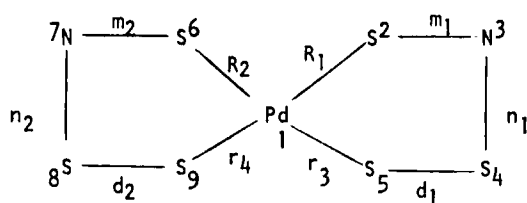
	Observed	$\Delta\nu^*$	Calculated	$\Delta\nu^*$	Predominant mode	Potential energy distribution(%)
Ag: ν_1	1262	0.0	1266.0	0.0	$\nu(\text{N}-\text{CH}_3)$	$S_6(90)$
(R) ν_2	1023	0.0	1026.5	0.0	$\nu(\text{S}_1-\text{N}_1)+\nu(\text{N}_1-\text{S}_2)$	$S_3(83)+S_5(12)$
ν_3	824	0.0	831.1	0.0	$\nu(\text{S}_2-\text{N}_2)+\delta(\text{S}-\text{N}-\text{CH}_3)$	$S_4(72)+S_{13}(14)$
ν_4	717	0.0	715.1	0.0	$\nu(\text{N}_1-\text{S}_2)+\delta(\text{S}-\text{N}-\text{S})$	$S_5(68)+S_{10}(18)$
ν_5	463	0.0	459.3	0.0	$\delta(\text{S}-\text{N}-\text{S})+\delta(\text{N}-\text{S}-\text{N})+$ $\delta(\text{Ni}-\text{N}-\text{S})+\delta(\text{S}-\text{N}-\text{CH}_3)$	$S_{10}(20)+S_{11}(28)+S_{12}(22)+S_{13}(11)$
ν_6	382	0.0	379.6	0.0	$\nu(\text{Ni}-\text{N})+\delta(\text{S}-\text{N}-\text{S})+\delta(\text{S}-\text{N}-\text{CH}_3)$	$S_2(18)+S_{10}(16)+S_{13}(44)$
ν_7	297	0.0	294.5	0.0	$\nu(\text{Ni}-\text{S})+\delta(\text{N}-\text{S}-\text{N})+$ $\delta(\text{Ni}-\text{N}-\text{S})+\delta(\text{S}-\text{N}-\text{CH}_3)$	$S_1(27)+S_{11}(13)+S_{12}(15)+S_{13}(27)$
ν_8	-	-	216.0	0.0	$\nu(\text{Ni}-\text{S})+\nu(\text{Ni}-\text{N})$	$S_1(43)+S_2(37)$
ν_9	-	-	156.6	0.0	$\nu(\text{Ni}-\text{S})+\nu(\text{Ni}-\text{N})+$ $\delta(\text{S}-\text{Ni}-\text{N})+\delta(\text{Ni}-\text{N}-\text{S})$	$S_1(11)+S_2(23)+S_8(32)+S_{12}(18)$

Bu:	ν_{10}	1268	0.0	1266.0	0.0	$\nu(\text{N-CH}_3)$	$\nu_{19}(90)$
(IR)	ν_{11}	1032	0.0	1026.6	0.0	$\nu(\text{S}_1-\text{N}_1)+\nu(\text{N}_1-\text{S}_2)$	$\nu_{16}(83)+\nu_{18}(12)$
	ν_{12}	834	0.0	831.6	0.0	$\nu(\text{S}_2-\text{N}_2)+\delta(\text{S-N-CH}_3)$	$\nu_{17}(72)+\nu_{26}(14)$
(IR)	ν_{13}	710	0.0	714.9	0.0	$\nu(\text{N}_1-\text{S}_2)+\delta(\text{S-N-S})$	$\nu_{18}(67)+\nu_{23}(18)$
	ν_{14}	465	0.0	466.3	0.45	$\nu(\text{Ni-N})+\delta(\text{S-N-S})+$ $\delta(\text{N-S-N})+\delta(\text{Ni-N-S})$	$\nu_{15}(10)+\nu_{23}(17)+\nu_{24}(25)+\nu_{25}(23)$
	ν_{15}	400	2.0	402.2	2.19	$\nu(\text{Ni-S})+\nu(\text{Ni-N})+$ $\delta(\text{S-N-S})+\delta(\text{S-N-CH}_3)$	$\nu_{14}(15)+\nu_{15}(37)+\nu_{23}(13)+\nu_{26}(24)$
	ν_{16}	321	3.0	323.0	4.40	$\nu(\text{Ni-S})+\nu(\text{Ni-N})$	$\nu_{14}(42)+\nu_{15}(35)$
	ν_{17}	310	1.0	311.3	1.25	$\nu(\text{Ni-S})+\delta(\text{N-S-N})+\delta(\text{S-N-CH}_3)$	$\nu_{14}(30)+\nu_{24}(12)+\nu_{26}(43)$
	ν_{18}	249	2.7	248.7	2.86	$\delta(\text{S-Ni-N})+\delta(\text{Ni-S-N})+$ $\delta(\text{S-N-S})+\delta(\text{Ni-N-S})$	$\nu_{20}(12)+\nu_{22}(10)+\nu_{23}(14)+\nu_{25}(36)$
	ν_{19}	-	-	90.4	0.61	$\delta(\text{S-Ni-N})+\delta(\text{S-Ni-N})$	$\nu_{20}(54)+\nu_{21}(38)$

$\Delta\nu^\# = \Delta(^{\text{NA}}\text{Ni} - ^{62}\text{Ni})$ NA - natural abundance

one normal modes are classified as $8 A_1 + 3 A_2 + 3 B_1 + 7 B_2$ and all modes are infrared and Raman active except A_2 which is only Raman active. In this treatment only the in-plane (A_1 and B_2) modes were considered and these twenty-two internal coordinates are shown in Table 5. The twenty-one symmetry coordinates as shown in Table 6 include five redundancies which were not easily

TABLE 5
Internal Coordinates for $\text{Pd}(\text{S}_3\text{N})_2$



$$2-1-6 = \pi_1$$

$$5-1-9 = \pi_2$$

$$2-1-5 = \theta_1$$

$$6-1-9 = \theta_2$$

$$1-2-3 = \delta_1$$

$$1-6-7 = \delta_2$$

$$2-3-4 = \gamma_1$$

$$6-7-8 = \gamma_2$$

$$3-4-5 = \beta_1$$

$$7-8-9 = \beta_2$$

$$4-5-1 = \alpha_1$$

$$8-9-1 = \alpha_2$$

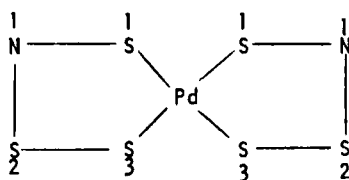
TABLE 6
Symmetry Coordinates for $\text{Pd}(\text{S}_3\text{N})_2$

$A_1:$	$S_1 = \Delta R_1 + \Delta R_2$	$S_6 = \Delta \Pi_1 + \Delta \Pi_2 - \Delta \theta_1 - \Delta \theta_2$
	$S_2 = \Delta r_3 + \Delta r_4$	$S_7 = \Delta \delta_1 + \Delta \delta_2$
	$S_3 = \Delta m_1 + \Delta m_2$	$S_8 = \Delta \gamma_1 + \Delta \gamma_2$
	$S_4 = \Delta n_1 + \Delta n_2$	$S_9 = \Delta \beta_1 + \Delta \beta_2$
	$S_5 = \Delta d_1 + \Delta d_2$	$S_{10} = \Delta \alpha_1 + \Delta \alpha_2$
$B_2:$	$S_{11} = \Delta R_1 - \Delta R_2$	$S_{17} = \Delta \Pi_1 - \Delta \Pi_2 - \Delta \theta_1 + \Delta \theta_2$
	$S_{12} = \Delta \gamma_3 - \Delta r_4$	$S_{18} = \Delta \delta_1 - \Delta \delta_2$
	$S_{13} = \Delta m_1 - \Delta m_2$	$S_{19} = \Delta \gamma_1 - \Delta \gamma_2$
	$S_{14} = \Delta n_1 - \Delta n_2$	$S_{20} = \Delta \beta_1 - \Delta \beta_2$
	$S_{15} = \Delta d_1 - \Delta d_2$	$S_{21} = \Delta \alpha_1 - \Delta \alpha_2$
	$S_{16} = \Delta \Pi_1 - \Delta \Pi_2 + \Delta \theta_1 - \Delta \theta_2$	

eliminated and were included in the calculations. These gave zero frequencies in the final results. The calculated force constants; stretching (K), bending (H) and repulsive (F) are shown in Table 7. The interaction force constants (f) are not shown. Table 8 shows the infrared and Raman spectra for $\text{Pd}(\text{S}_3\text{N})_2$, the predominant modes as well as the percent potential energy distribution (PED).

TABLE 7
^o
 Force Constants (mdy/A) for Pd(S₃N)₂

K(Pd-S ₁)	1.08	H(S-N-S)	0.73
K(Pd-S ₃)	0.98	H(N-S-S)	0.90
K(S ₁ -N ₁)	5.50	H(Pd-S-S)	0.30
K(N ₁ -S ₂)	2.68	F(S ₁ -S ₁)	0.76
K(S ₂ -S ₃)	1.52	F(S ₁ -S ₃)	0.29
H(S-Pd-S)	0.90	F(Pd-N)	0.10
H(S ₁ -Pd-S ₃)	0.90	F(S ₁ -S ₂)	0.42
H(Pd-S-N)	0.20	F(N-S)	0.20
		F(Pd-S)	0.40



RESULTS AND DISCUSSION

Replacement of the N-H hydrogens of *cis*-Ni(S₂N₂H)₂ with methyl groups results in a *trans*-Ni(S₂N₂CH₃)₂ (See figure in Table 3). Two of the three ring S-N stretching modes are similar to those found in *cis*-Ni(S₂N₂H)₂: $\nu(\text{S}_1\text{-N}_1)$, 1032 cm⁻¹ and

TABLE 8

Comparison of Observed and Calculated Frequencies and Band Assignments for Pd(S ₃ N) ₂ (cm ⁻¹)					
	Observed		Calculated	Predominant mode	Potential energy distribution (%)
A ₁ : (R)	ν ₁	1060	1059.9	ν(S ₁ -N ₁)	S ₃ (81)
	ν ₂	700	701.7	ν(N ₁ -S ₂)	S ₄ (68)
	ν ₃	515	512.4	ν(S-S)+δ(S-Pd-S)+δ(N-S-S)	S ₅ (45)+S ₆ (15)+S ₉ (16)
	ν ₄	444	444.0	ν(S-S)+δ(S-N-S)+δ(N-S-S)+δ(Pd-S-S)	S ₅ (25)+S ₈ (19)+S ₉ (14)+S ₁₀ (11)
	ν ₅	350	353.3	ν(Pd-S ₃)+δ(S-Pd-S)+δ(Pd-S-S)	S ₂ (54)+S ₆ (10)+S ₁₀ (13)
	ν ₆	335	335.7	ν(Pd-S ₁)+ν(Pd-S ₃)	S ₁ (40)+S ₂ (40)
	ν ₇	220	220.6	ν(Pd-S ₁)+ν(Pd-S ₃)	S ₁ (35)+S ₂ (55)
	ν ₈	-	9.4	δ(Pd-S-N)+δ(Pd-S-S)	S ₇ (29)+S ₁₀ (54)
B ₂ : (IR)	ν ₉	1050	1050.6	ν(S ₁ -N ₁)	S ₁₃ (81)
	ν ₁₀	700	698.3	ν(N ₁ -S ₂)	S ₁₄ (67)
	ν ₁₁	474	476.2	ν(N ₁ -S ₂)+ν(S-S)+δ(S-N-S)+δ(N-S-S)	S ₁₄ (11)+S ₁₅ (36)+S ₁₉ (14)+S ₂₀ (24)
	ν ₁₂	394	396.6	ν(S-S)+δ(S-N-S)+δ(Pd-S-S)	S ₁₅ (55)+S ₁₉ (13)+S ₂₁ (12)
	ν ₁₃	354	351.5	ν(Pd-S ₁)+ν(Pd-S ₃)	S ₁₁ (69)+S ₁₂ (22)
	ν ₁₄	272	269.3	ν(Pd-S ₁)+δ(S-Pd-S)+δ(S ₁ -Pd-S ₃)+δ(Pd-S-S)	S ₁₁ (11)+S ₁₆ (28)+S ₁₇ (28)+S ₂₁ (18)
	ν ₁₅	248	250.1	ν(Pd-S ₃)+δ(S-Pd-S)	S ₁₂ (13)+S ₁₆ (49)
	ν ₁₆	191	189.6	ν(S-S)+δ(S-Pd-S)+δ(S ₁ -Pd-S ₃)	S ₁₅ (22)+S ₁₆ (33)+S ₁₇ (33)

$\nu(\text{N}_1\text{-S}_2)$, 717 cm^{-1} . The $\nu(\text{N}_2\text{-S}_2)$ band is shifted from 890 cm^{-1} in *cis*- $\text{Ni}(\text{S}_2\text{N}_2\text{H})_2$ to 834 cm^{-1} in *trans*- $\text{Ni}(\text{S}_2\text{N}_2\text{CH}_3)_2$. Most bands in *trans*- $\text{Ni}(\text{S}_2\text{N}_2\text{CH}_3)_2$ as well as in *cis*- $\text{Ni}(\text{S}_2\text{N}_2\text{H})_2$ are strongly coupled. The metal isotope substitution $^{64}\text{Ni}/^{62}\text{Ni}$ is expected to shift all the metal-nitrogen and metal-sulfur stretching modes.

It is particularly significant that the calculations due to ^{62}Ni isotope substitution are in good agreement with those observed for the Ni-N and Ni-S stretching and skeletal bending modes (Table 4). Band assignments for the observed frequencies and isotopic shift are in good agreement and show an average error of less than one percent.

For planar $\text{Pd}(\text{S}_3\text{N})_2$ (See figure in Table 7) the ring S-N assignments can be compared to our previous synthesis and infrared study of $\text{Pt}(\text{S}_3\text{N})_2$ (2). In the platinum complex $\nu(\text{S}_1\text{-N}_1)$ is shifted from 1025 cm^{-1} to 1050 cm^{-1} in the palladium complex. The $\nu(\text{N}_1\text{-S}_2)$ vibration shows a small change shift, 712 cm^{-1} for platinum to 700 cm^{-1} for the palladium complex. While no NCA was carried out on $\text{Pt}(\text{S}_3\text{N})_2$, the $\text{Pd}(\text{S}_3\text{N})_2$ complex shows strong coupling for most of the stretching and bending modes. Band assignments show an average error of 0.05 percent.

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